

Upscaling of Mixing-controlled Reactive Transport

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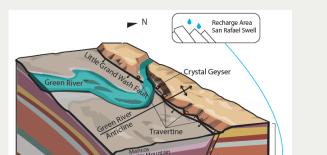


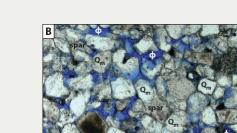
We combine laboratory micro-scale experimental and modeling efforts to examine:

- How does scCO₂ interact with brines and mineral surfaces? [multiphase flow]
- What are the relevant physics of dissolved CO₂ transport? [reactive transport and rock-fluid interactions]

3D FI	ow Field in	n a mi	cromo	del
	Experimenta	al setup		
D Laser scanning co	onfocal microscope			
		Lateral view of micromodel	Optical sections	2D slices of 3D field

Natural Analogue for Carbonate Sealing of CO₂ Leakage Pathways

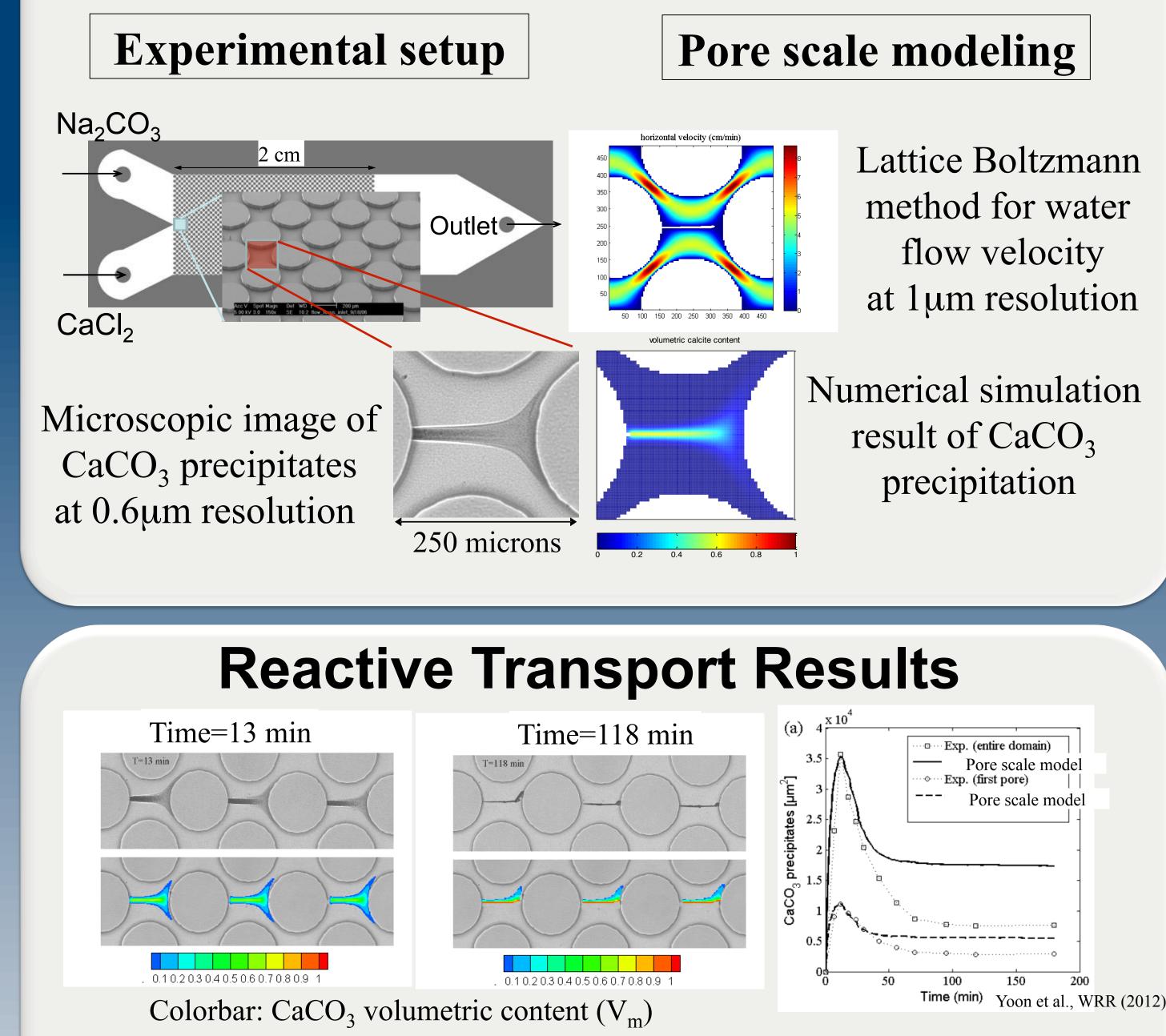


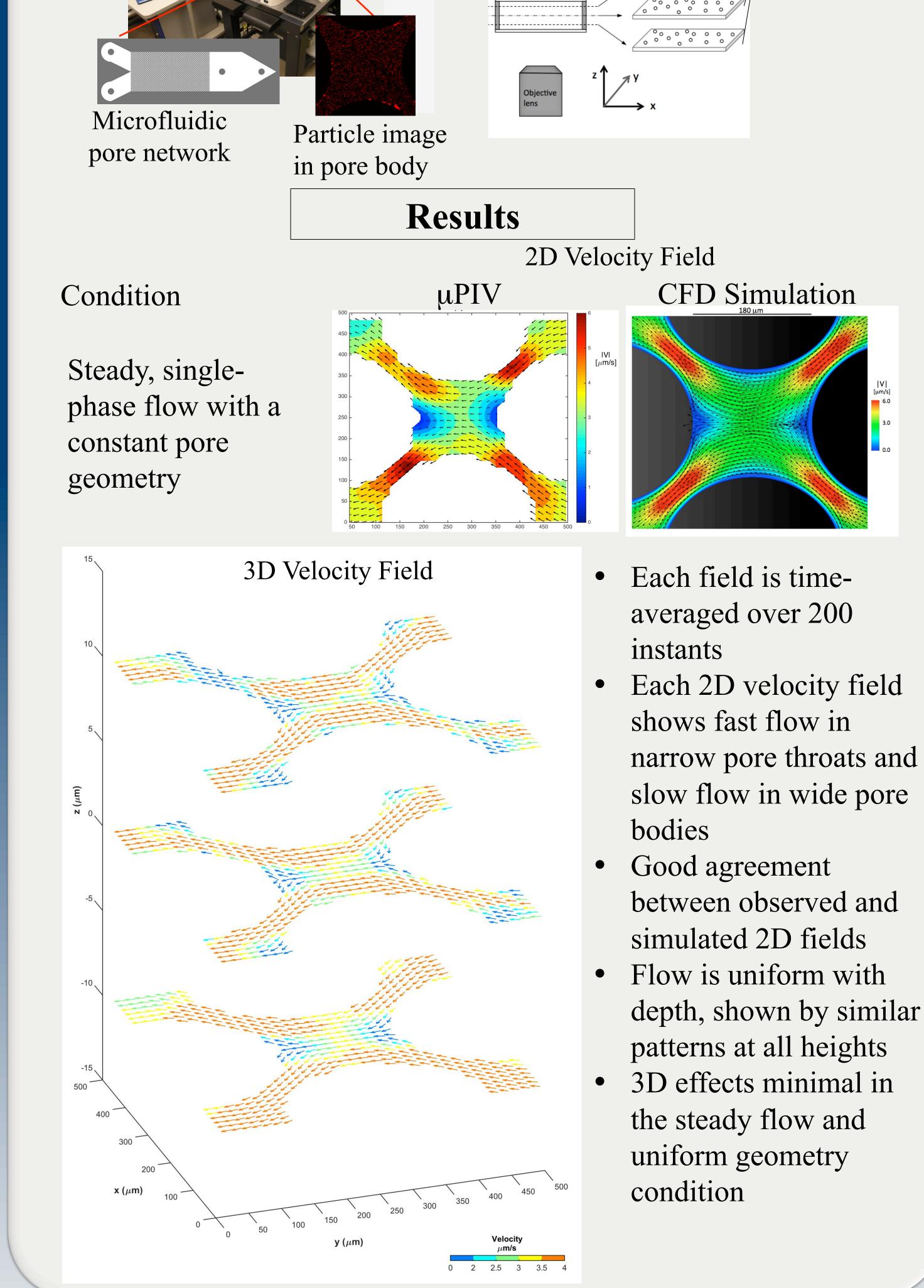


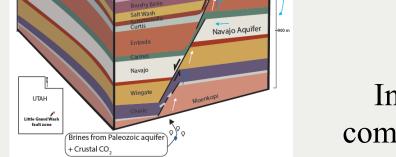
• How can pore scale processes be synthesized and upscaled into more powerful continuum models? [upscaling]

Pore Scale Modeling for Reactive Transport

• Transverse mixing-induced calcium carbonate (CaCO₃) precipitation







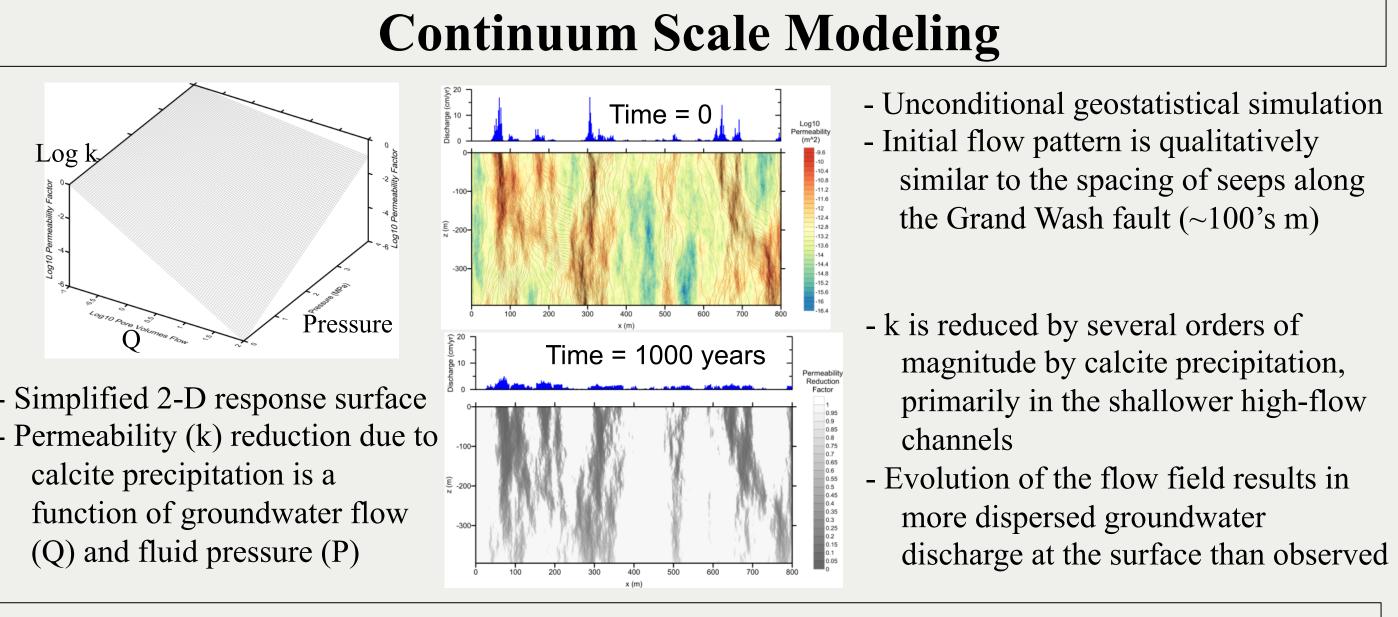


Inside fault zone: Porosity Outside fault zone: High porosity completely filled with calcite spar and absence of calcite cement

Little Grand Wash Fault, Crystal Geyser, Utah

Courtesy of P. Eichhubl

•Observations along the surface exposure of the Grand Wash fault indicate alteration zones of 10-50 m width with spacing on the order of 100 m •Locations of conduits controlled by fault-segment intersections and topography •Sandstone permeability reduced by 3 to 4 orders of magnitude in alteration zones by carbonate cementation



Response Function based on Pore Scale Simulations

• Phenomenological power law relations for continuum scale model can be derived from

• Simulations capture precipitation & dissolution patterns observed in the micromodel • A decrease of precipitate area (dissolution) was captured by using a dissolution factor which accounts for high surface area of nano-particles, transformation to different forms of CaCO₃, and stability of nano-particles after pore blocking as shown below





Calcium Carbonate: Polymorphs $[Ca^{2+}]_T = [CO_3^{2-}]_T = 10 \text{mM} \& [Mg^{2+}]_T = 40 \text{ mM}$ $[Ca^{2+}]_{T} = [CO_{3}^{2-}]_{T} = 6.5 \text{ mM}$ opic image at 8 hrs

slow flow in wide pore

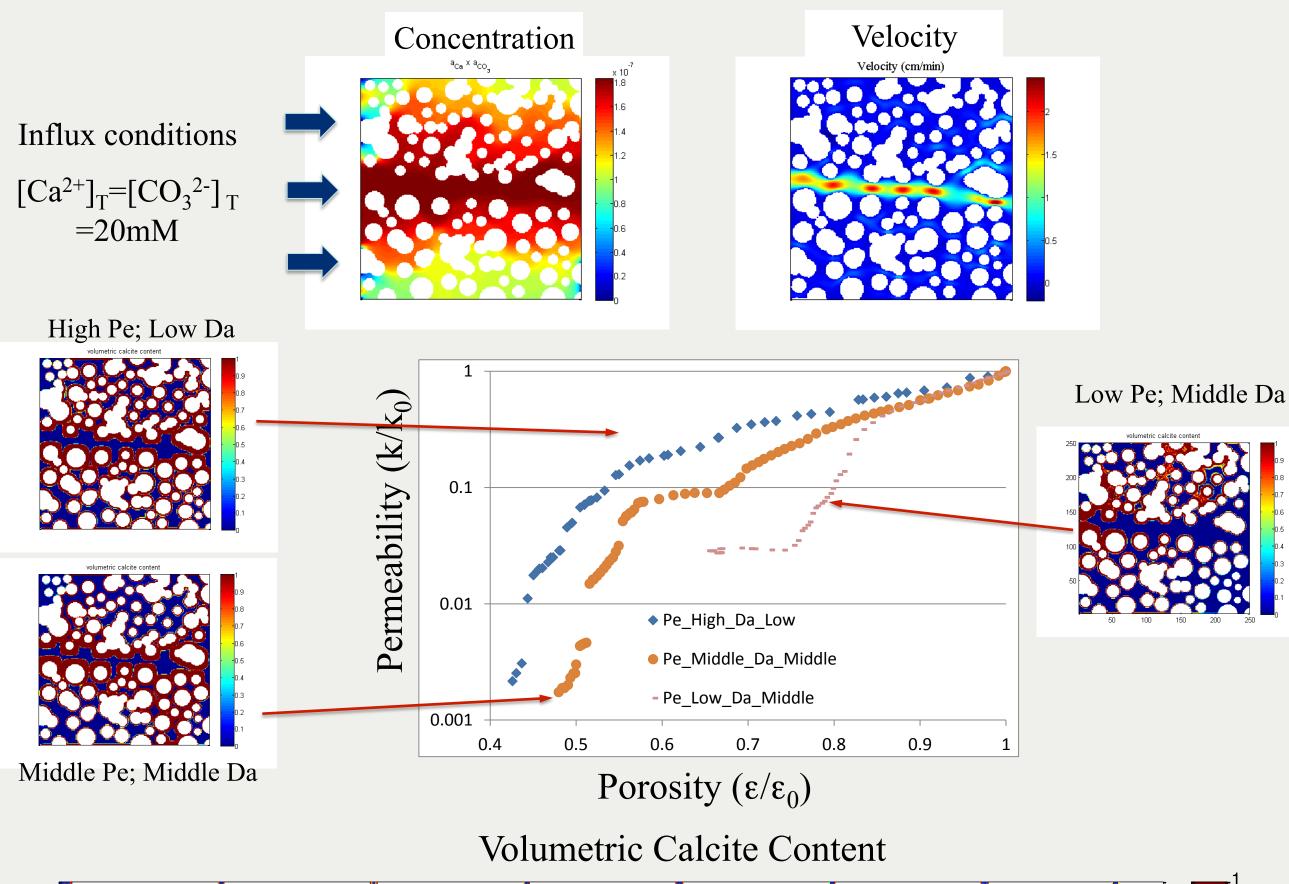
=20mM

- between observed and
- Flow is uniform with depth, shown by similar patterns at all heights 3D effects minimal in

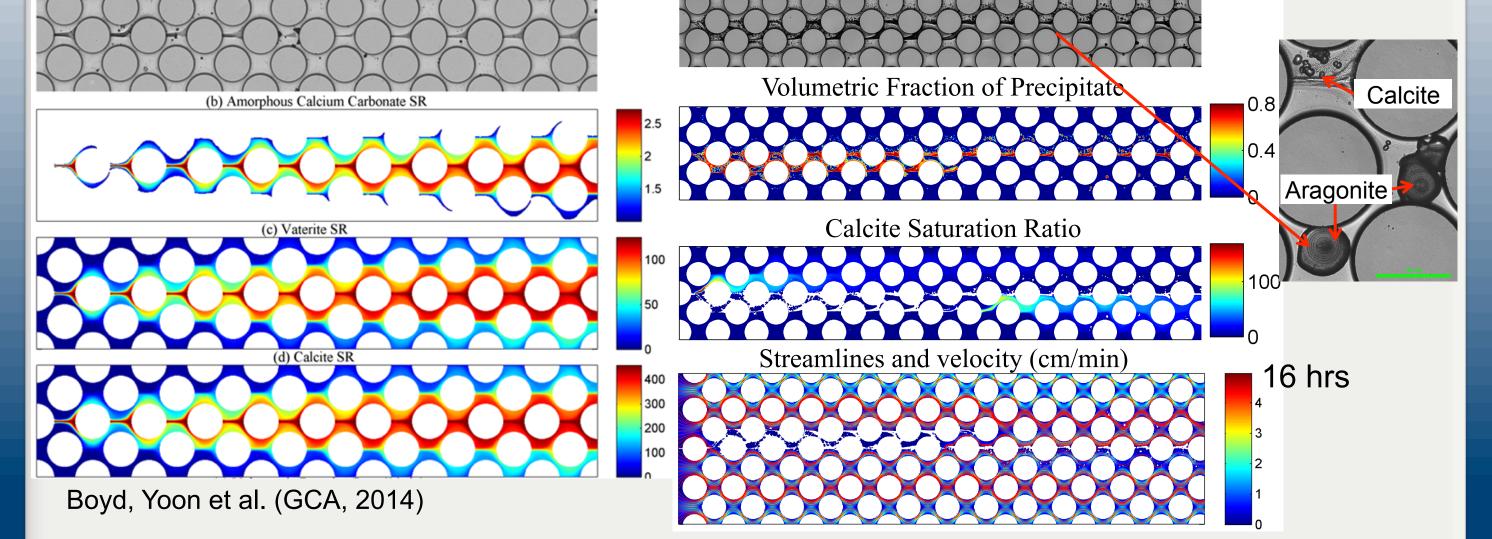
Simulated flow in a micromodel

- Mixing-induced chemical reactions can alter fluid properties (viscosity and density), mixing efficiency, and shear rate for engineered solutions Product conc.

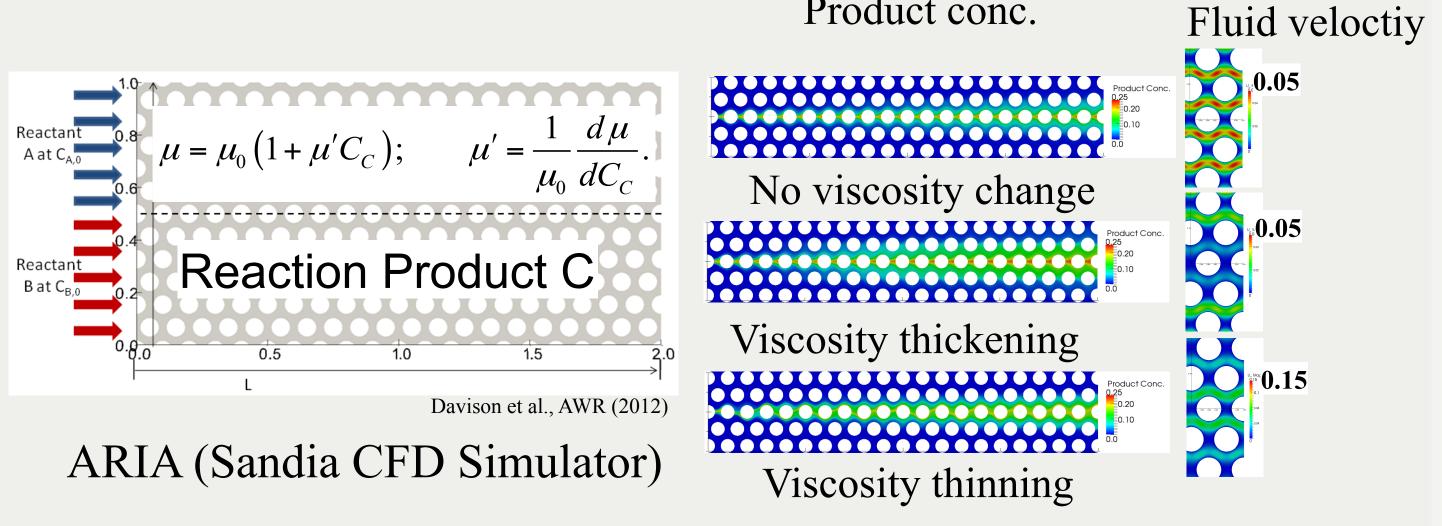
pore scale simulations (permeability-porosity, tortuosity-porosity, surface area-porosity)







- Mineral precipitation rate along flow direction is concentration dependent and limited by transverse mixing
- CaCO₃ mineral phases (i.e., polymorphs) are chemistry dependent and affected by flow and kinetics
- Overall, reaction kinetics, crystal growth and morphology are spatially and temporally affected by solution chemistry and hydrodynamics at pore scale • Pore-scale model can be used to test if pore-scale processes observed in micromodels is predicted, and to develop an upscaled reaction model



- Pore scale model with high performance computing capability was used to test reactive transport experimental results under a variety of pore-geometry conditions
- Same model will be used to test flow experimental results under a variety of pore-geometry conditions
- Simulator reproduces flow experimental results under steady and constant geometry condition

Influx conditions: $[Ca^{2+}]_T = [CO_3^{2-}]_T = 20 \text{mM}$

Summary and Implications

• Vigorously tested pore-scale model can be used to develop a response function (or dimension reduction model) for continuum-scale relationships

• k-ε and surface area-ε relationships will be developed over a range of solution chemistry, chemical reaction, and pore structure configurations • An adaptive strategy to couple continuum and pore-scale using a response function approach as well as hybrid pore-continuum model using p-Flotran) will be tested

Selected Publications

• Davison, S. M., H. Yoon, M. J. Martinez (2012), Pore scale analysis of the impact of mixing-induced reaction dependent viscosity variations, Advances in Water Resources, 38, 70-80.

- Yoon, H., A. J.Valocchi, C. J.Werth, and T.Dewers (2012), Pore-scale simulation of mixing-induced calcium carbonate precipitation and dissolution in a microfluidic pore network, Water Resour. Res., 48, W02524, doi: 10.1029/2011WR011192.
- Boyd T, Yoon H, Zhang C, Dehoff K, Fouke B, Valocchi AJ, Werth CJ (2014) Infl uence of Mg2+ on CaCO3 precipitation during subsurface reactive transport in a homogeneous silicon-etched pore network. Geochim Cosmochim Acta 135:321–335
- Martinez MJ, Stone CM, Notz PK, Turner DZ, Hopkins PL, Subia S, et al. Computational thermal, chemical, fluid, and solid mechanics for geosystems management. Technical Report, SAND2011-6643, Sandia National Laboratories, Albuquerque, NM 2011.

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Reactant